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(Chapter 6) :

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In a reaction,

5.3 g of sodium

carbonate

reacted with 6 g

of acetic acid.

The products

were 2.2 ...

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list using the

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Cover, Write,

Check' activity.

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using the word

list: acids and

alkalis. Test

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activities:
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And Hydrochloric

acid, citric

acid, acetic

acid, nitric

acid, formic

acid, sulphuric

acid. Answer.

Strong acids are

those acids

which are

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completely
ionised in

aqueous

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solution. Weak

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which do not

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2 Acids, bases
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acid, citric
acid, acetic
acid, nitric

acid, formic
acid, sulphuric
acid. Answer.

Strong acids are those acids which are completely ionised in aqueous solution.

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Class 7 Science
Chapter 5 Acid
Base and ...
Acids form
aqueous
solutions with a
sour taste, can

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turn blue litmus

red, and react

with bases and

certain metals

(like calcium)

to form salts.

The word acid is

derived from the

Latin

acidus/acēre,

meaning 'sour'.

An aqueous

solution of an

acid has a pH

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less than 7 and
is

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2. Circle the
correct bold
word in the
sentences below.
Acids taste
sour/feel soapy.

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Some acids and alkalis are corrosive/correlated . A concentrated solution of an acid is more/less corrosive than a dilute solution. A concentrated solution has fewer/more acid particles per litre than a

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dilute solution.

And Bases

Science Home

Learning Task

*Year 8 Acids and
alkalis*

Reactions of
acids with
metals. Acids
react with most
metals and, when
they do, a salt
is produced. But
unlike the

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reactions between

acids and bases,

we do not get

water.

Type

Reactions of

acids with

metals - Metals

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other devices

with screenshots

for you to solve

the levels

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an acid

(hydrochloric,

sulfuric or

nitric) of

unknown

concentration;

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sodium hydroxide

solution of

known

concentration; a

suitable

indicator, for

example methyl

orange or

phenolphthalein

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Type base can be
represented by
the general word
equation shown
below: acid +
base \rightarrow salt +
water . A base
is a substance
that contains

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hydroxide (1) A
base is often a
compound made up
of a metal and
hydroxide.

Common bases
are: sodium ...

Acid and Base

Word Equations

Chemistry

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And Bases
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Acidic solutions
have pH values
lower than 7.

Acid-alkali
indicators can
show whether a
solution is
acidic, neutral
(pH 7) or
alkaline. The

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table shows the

colours for two

different common

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Acidic and

alkaline

solutions -

Acids and bases

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Quiz on Acids
and Alkalis for

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crystals form

when the hot

saturated

solution is

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Type Answer. A.

Because the
solubility of
copper(iii)
sulfate

increases with
temperature. B.

Because the
solubility of

...

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Alkalis Quiz -
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Q5: In
neutralisation
reactions
between an acid
and alkali.
Hydrogen ions
react with
hydroxide ions
to produce
water. Represent

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this in an Acids
equation. (1
mark) Strong and
weak acids Q6:

Complete the
sentences using
the words in the
boxes. A acid is
ionised in
aqueous
solution. A weak
acid is ionised
in aqueous
solution. (3

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marks) 1 17 7 4

Solutions Acids And Bases

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Science

Many acids and bases occur naturally in nature, such as citric acid in fruits like orange, lemon etc, tartaric acid in

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tamarind, malic acid in apples and lactic acid in milk and milk products, hydrochloric acid in gastric juices.

Similarly, many bases are found such as lime water. We use many of these acids in our day-

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to-day life, such as vinegar or acetic acid in the kitchen, boric acid for laundry, baking soda for the purpose of cooking, washing soda for cleaning etc.

*Acids, Bases,
and Salts -
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Introduction, Acids

Dissociation ...

Acids form
acidic solutions

in water. Acids
produce hydrogen

ions, H^+ in
aqueous

solution. For
example: HCl

$(aq) \rightarrow H^+(aq) +$
 $Cl^-(aq)$ Acidic

solutions have
pH values less

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Action helps
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between the
science they

read and what

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content, lively

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students'

understanding of

science beyond

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support
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Solutions for
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Chemistry has
been divided
into 3 parts.

Part A provides
detailed
solutions (Question-by-Question)
of all the
questions/
exercises

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Part B provides
solutions to the
questions in the
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book. Part C
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selected
Practice
Questions useful
for the Class 11
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along with

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face JEE Main
competitive
examination 9.
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students to face
the competitive
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a supreme state

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and ensures
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